



## Chemistry Scope and Sequence

	Quarter: 1	Quarter2	Quarter 3	Quarter 4
	<b>STRUCTURE AND PROPERTIES OF MATTER</b>	<b>STRUCTURE AND PROPERTIES OF MATTER</b>	<b>INTERACTIONS OF MATTER</b>	<b>INTERACTIONS OF MATTER</b>
<b>Content</b>	<p><b>C.PM.1:</b> Atomic structure</p> <ul style="list-style-type: none"> <li>• Evolution of atomic models/theory</li> <li>• Electrons</li> <li>• Electron configurations</li> </ul> <p><b>C.PM.2:</b> Periodic Table</p> <ul style="list-style-type: none"> <li>• Properties</li> <li>• Trends</li> </ul> <p><b>C.PM.3:</b> Chemical bonding</p> <ul style="list-style-type: none"> <li>• Ionic</li> <li>• Polar/covalent</li> </ul>	<p><b>C.PM.4:</b> Representing compounds</p> <ul style="list-style-type: none"> <li>• Formula writing</li> <li>• Nomenclature</li> <li>• Models and shapes (Lewis structures, ball and stick, molecular geometries)</li> </ul> <p><b>C.PM.5:</b> Quantifying matter</p> <p><b>C.PM.6:</b> Intermolecular forces of attraction</p> <ul style="list-style-type: none"> <li>• Types and strengths</li> <li>• Implications for properties of substances</li> <li>▪ Melting and boiling point</li> <li>▪ Solubility</li> <li>▪ Vapor pressure</li> </ul>	<p><b>C.IM.1:</b> Chemical reactions</p> <ul style="list-style-type: none"> <li>• Types of reactions</li> <li>• Kinetics</li> <li>• Energy</li> <li>• Equilibrium</li> <li>• Acids/bases</li> </ul> <p><b>C.IM.2:</b> Gas laws</p> <ul style="list-style-type: none"> <li>• Pressure, volume and temperature</li> <li>• Ideal gas law</li> </ul>	<p><b>C.IM.3:</b> Stoichiometry</p> <ul style="list-style-type: none"> <li>• Molecular calculations</li> <li>• Solutions</li> <li>• Limiting reagents</li> </ul>
<b>Resources</b>	<b>McGraw Hill ODE Model Curriculum</b>	<b>McGraw Hill ODE Model Curriculum</b>	<b>McGraw Hill ODE Model Curriculum</b>	<b>McGraw Hill ODE Model Curriculum</b>
<b>Vocabulary</b>				

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**Quarter(s) 1-4**

**During the years of 9-12, all students must become proficient in the use of the following scientific processes, with appropriate laboratory safety techniques, to construct their knowledge and understanding in all science content areas:**

**SIA1** Identify questions and concepts that guide scientific investigations.

**SIA2** Design and conduct scientific investigations.

**SIA3** Use technology and mathematics to improve investigations and communications.

**SIA4** Formulate and revise explanations and models using logic and evidence (critical thinking).

**SIA5** Recognize and analyze explanations and models.

**SIA6** Communicate and support a scientific argument.

**WHCSD Scope and Sequence**

**Chemistry**

**2021-2022**